

Università di Torino

NIS Colloquium

Classical and new approaches to thin film photovoltaics

Monday, June 23, 2008

Aula Magna, Experimental Physics Department, Via Pietro Giuria 1, Torino

Dye Sensitized Solar Cells

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Dipartimento di Chimica Generale e Chimica Organica
Centro Interdipartimentale di Eccellenza NIS (Superfici ed Interfasi Nanostrutturate)
Università degli Studi di Torino

ENERGY FROM THE SUN



Energy needs: more than **15TW** per year

Sun energy on earth surface: **120000 TW**



Energy delivered from the sun in Italy (centre): 1400 kWh/m² per year

Family needs: 2kWp (approximately)

16-20 m² of photovoltaic devices with a 12.5-10% efficiency

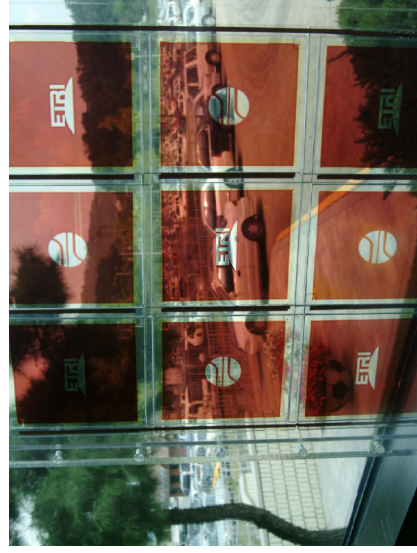
Lower the photovoltaic module cost per Wp/m²!

C. Barolo – Dye Sensitized Solar Cells

DSSC ADVANTAGES

 *Low cost material*

 *Processes: simple and low cost*



*Low cost,
different colors*



*Transparent, light weight
Flexibility
Easy of integration*



Efficiency independent from temperature



Can use diffuse light!

1991: Nanocrystalline photovoltaic cells

LOW COST HIGH EFFICIENCY SOLAR CELLS



LETTERS TO NATURE

A low-cost, high-efficiency solar cell based on dye-sensitized colloidal TiO_2 films

Brian O'Regan* & Michael Grätzel†

Institute of Physical Chemistry, Swiss Federal Institute of Technology, CH-1015 Lausanne, Switzerland

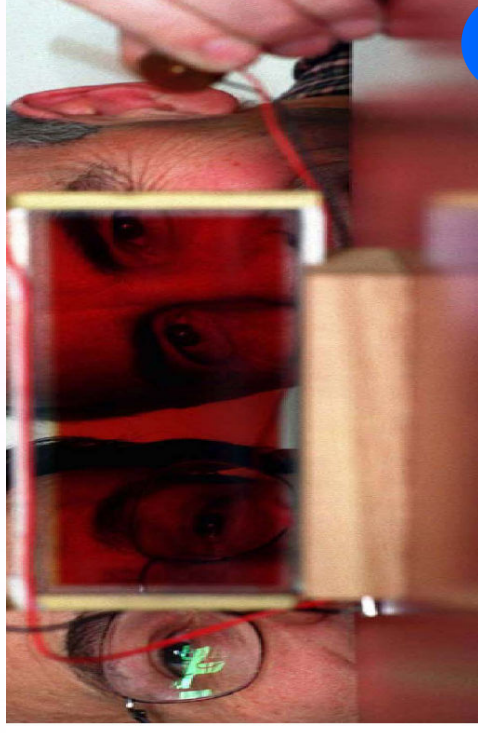
THE large-scale use of photovoltaic devices for electricity generation is prohibitively expensive at present: generation from existing commercial devices costs about ten times more than conventional methods¹. Here we describe a photovoltaic cell, created from low-to medium-purity materials through low-cost processes, which exhibits a commercially realistic energy-conversion efficiency. The device is based on a 10- μm -thick, optically transparent film of titanium dioxide particles a few nanometres in size, coated with a monolayer of a charge-transfer dye to sensitize the film for light harvesting. Because of the high surface area of the semiconductor film and the ideal spectral characteristics of the dye, the device harvests a high proportion of the incident solar energy flux (46%) and shows exceptionally high efficiencies for the conversion of incident photons to electrical current (more than 80%). The overall light-to-electric energy conversion yield is 7.1–7.9% in simulated solar light and 12% in diffuse daylight. The large current densities (greater than 12 mA cm^{-2}) and exceptional stability (sustaining at least five million turnovers without decomposition), as well as the low cost, make practical applications feasible.

*Present address: Department of Chemistry, University of Washington, Seattle, Washington 98195, USA.

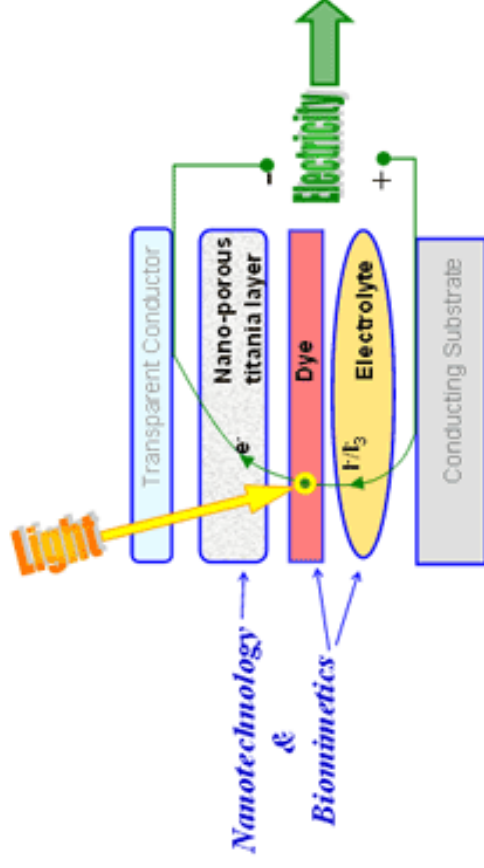
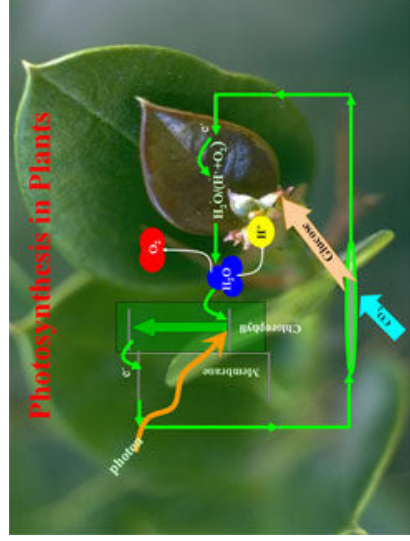
† To whom correspondence should be addressed.

NATURE • VOL 353 • 24 OCTOBER 1991

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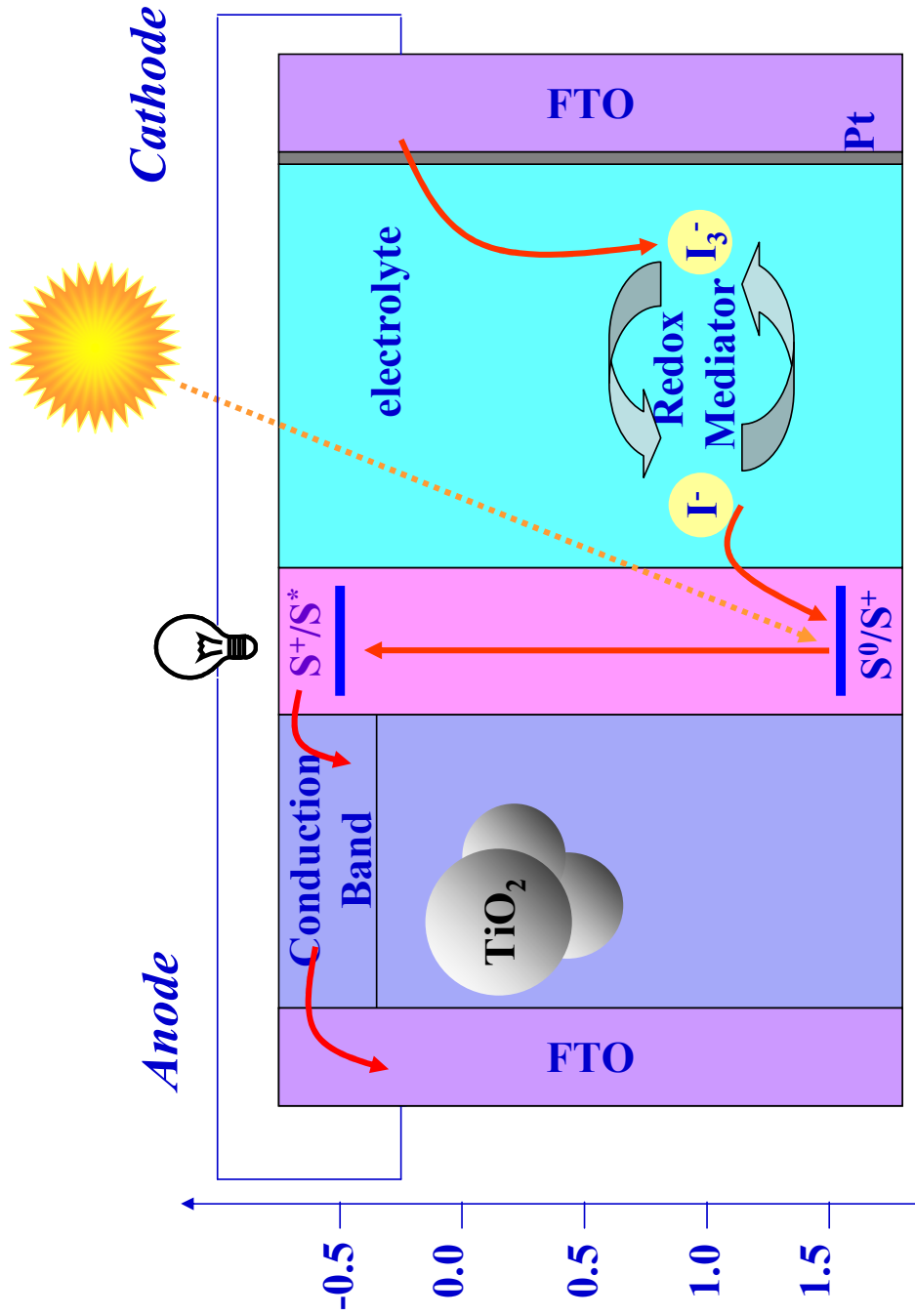
POWER FROM ARTIFICIAL PHOTOSYNTHESIS



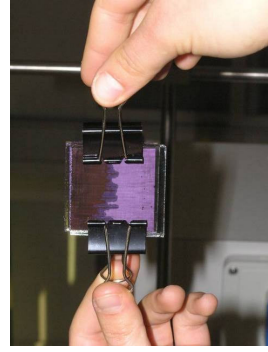
Copyright © 2005-2008, Greatcell. All rights reserved.

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DSC: how does it work?



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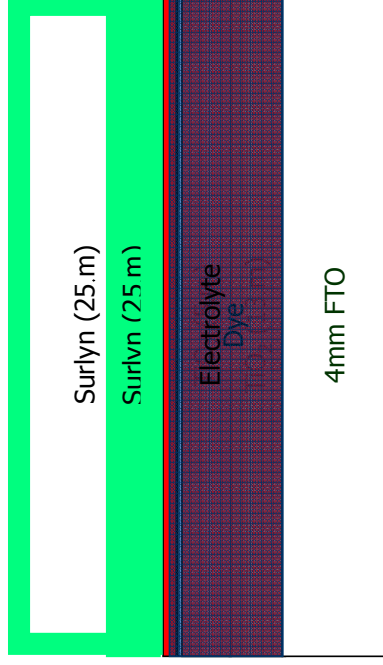
DSC's: cell preparation



2.3mm FTO

Pt

Tenda della scienza 9-11 maggio 2008

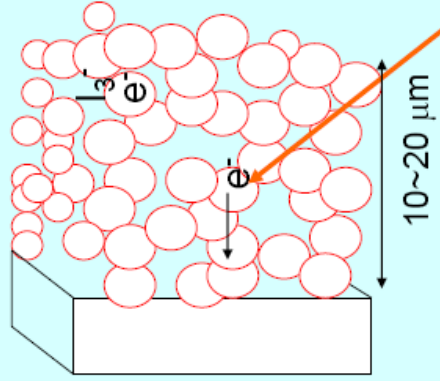


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DSC: how does it work?

Dye sensitized nanocrystals show quantitative conversion of the photons into electric current

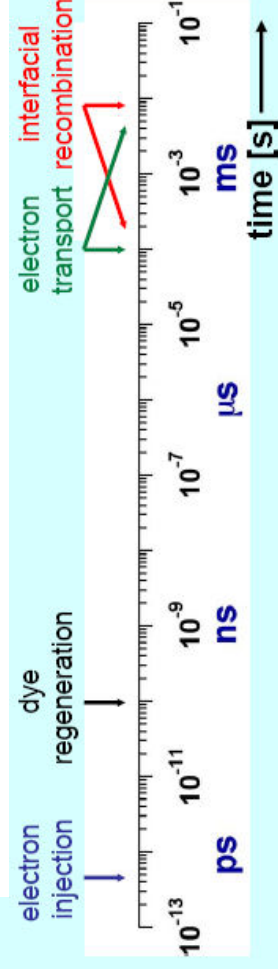
SnO₂:F TiO₂/Electrolyte



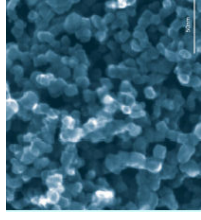
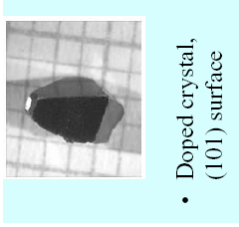
Electrons should travel to the SnO₂ before charge recombination occurs

Diffusion length should exceed the thickness of the mesoscopic TiO₂ film

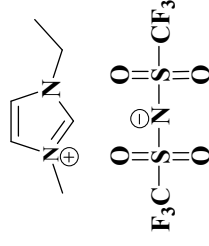
DYNAMIC COMPETITION



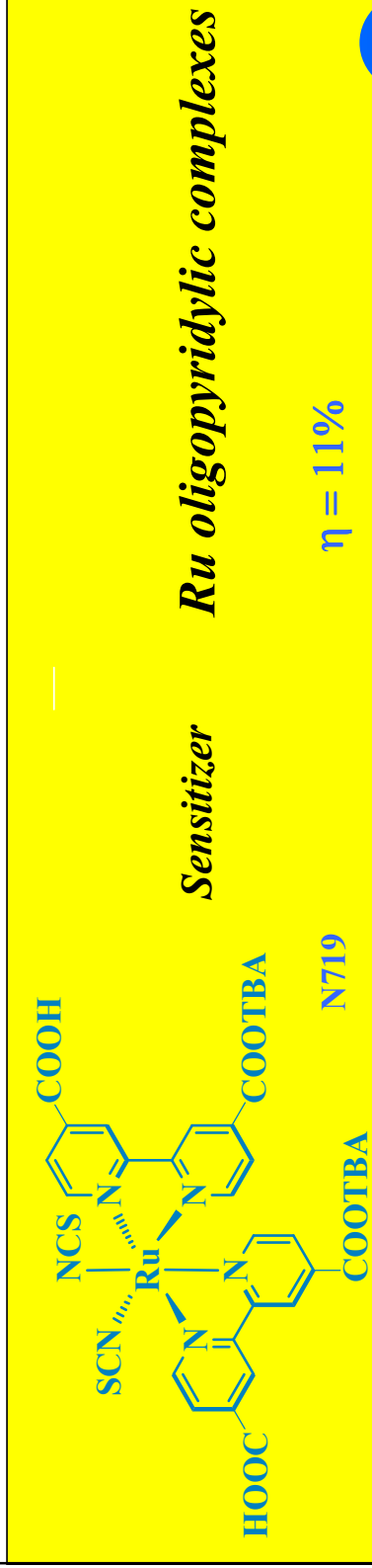
DSSC: components



Wide band gap semiconductor TiO_2 anatase



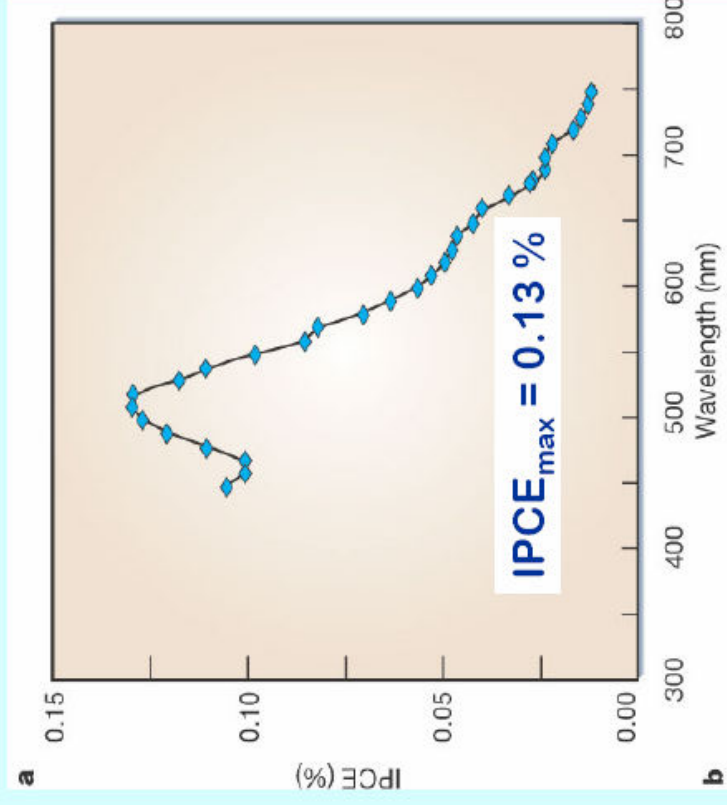
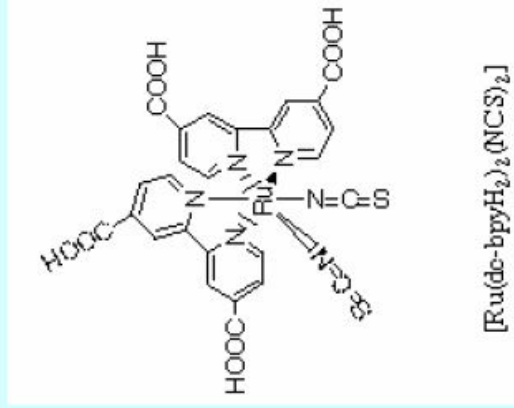
Robust electrolyte ionic liquid



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Nanostructured TiO_2

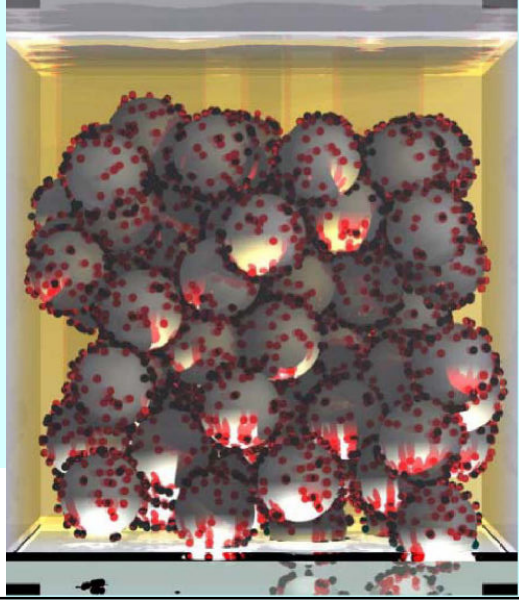
Incident photon to electron conversion efficiency (IPCE) of a dye-sensitized TiO_2 (101) single crystal PEC solar cell



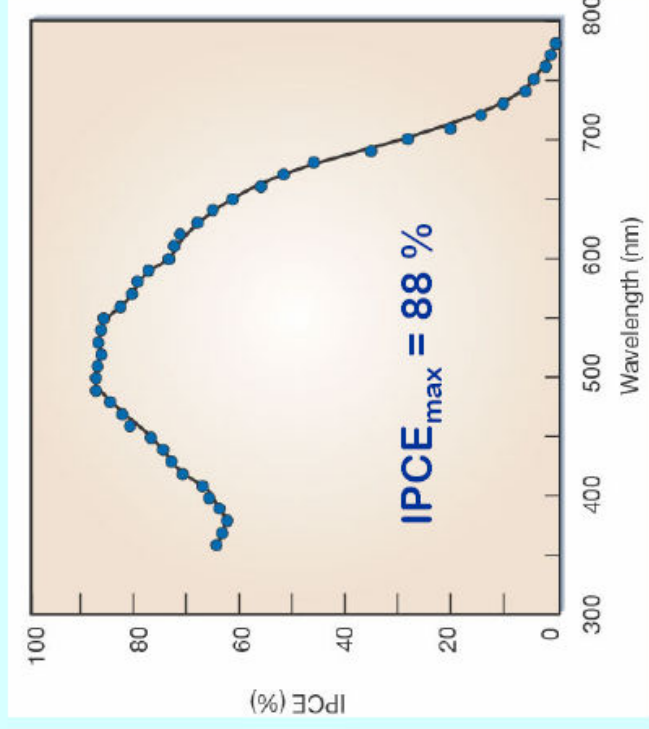
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Nanostructured TiO_2

Incident photon to current conversion efficiency of a dye-sensitized solar cell based on a mesoscopic TiO_2 electrode



Courtesy of Dr. Arthur J. Frank, NREL, USA



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TiO₂ paste

FABRICATION OF SCREEN-PRINTING PASTES

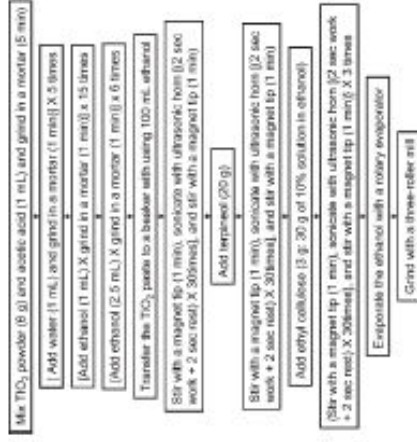


Figure 1. Fabrication scheme of screen-printing paste from a nanocrystalline-TiO₂ powder

FABRICATION OF SCREEN-PRINTING PASTES

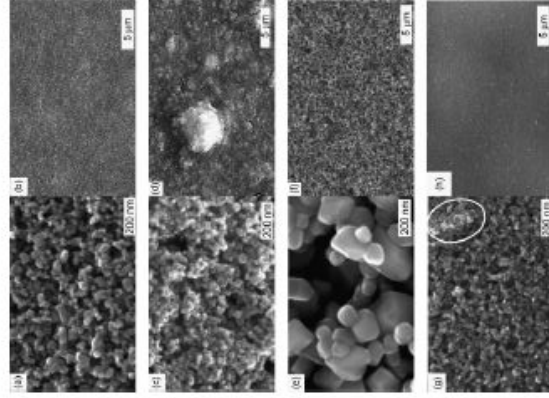


Figure 2. SEM photographs of porous TiO₂ films showing the surface morphology of P25 (a, b), ST21 (c, d), ST41 (e, f) and homemade TiO₂ (g, h) with two types of magnification: 120000× (a, c, e, g) and 5000× (b, d, f, h)

S. Ito et al. Prog. Photovolt: Res. Appl. 2007
DOI: 10.1002/pip

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ELECTROLYTE

SOLVENT FREE SYSTEM:

- solid electrolytes (polymer)
- solidified ionic liquid
- hole conductors

ION-GEL Electrolyte (NEDO)

Features of Ionic Liquids

- Consists of only ions
- Liquid under wide temp. range ex. -10°C to 400°C
- non volatile
- Chemically stable and non combustible
- High electronic conductivity



1-Ethyl-3-methylimidazolium - Bis(trifluoromethylsulfonyl) Amide

EMIm-TFSA

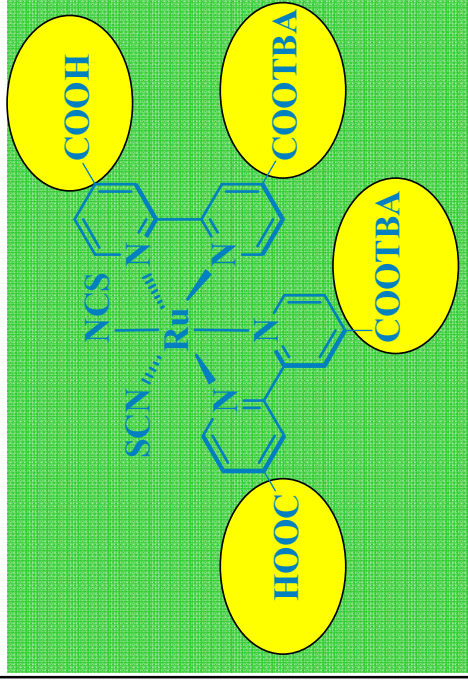
Sensitizers Engineering Approaches

Sensitizers Synthesis

Outlines

- Classical Organometallic Sensitizers
- Tetradentate Organometallic Sensitizers
- Organic dyes: a new generation of sensitizers?
- *Characterization of the sensitizer absorbing solution*

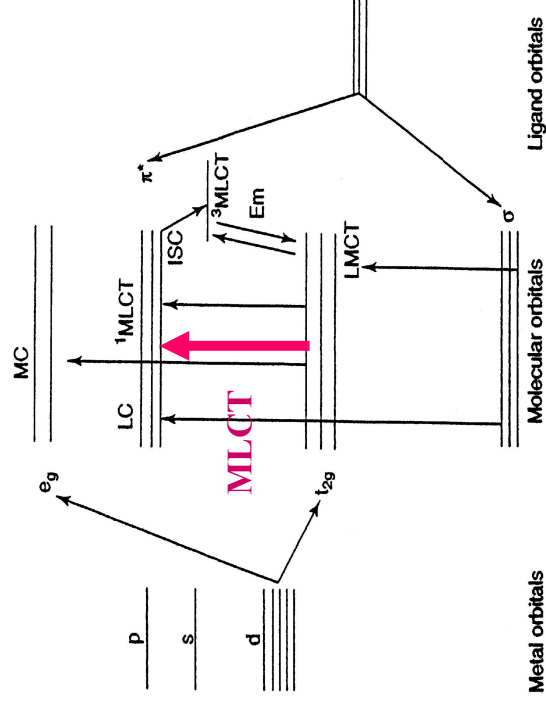
Sensitizers: Engineering Approaches



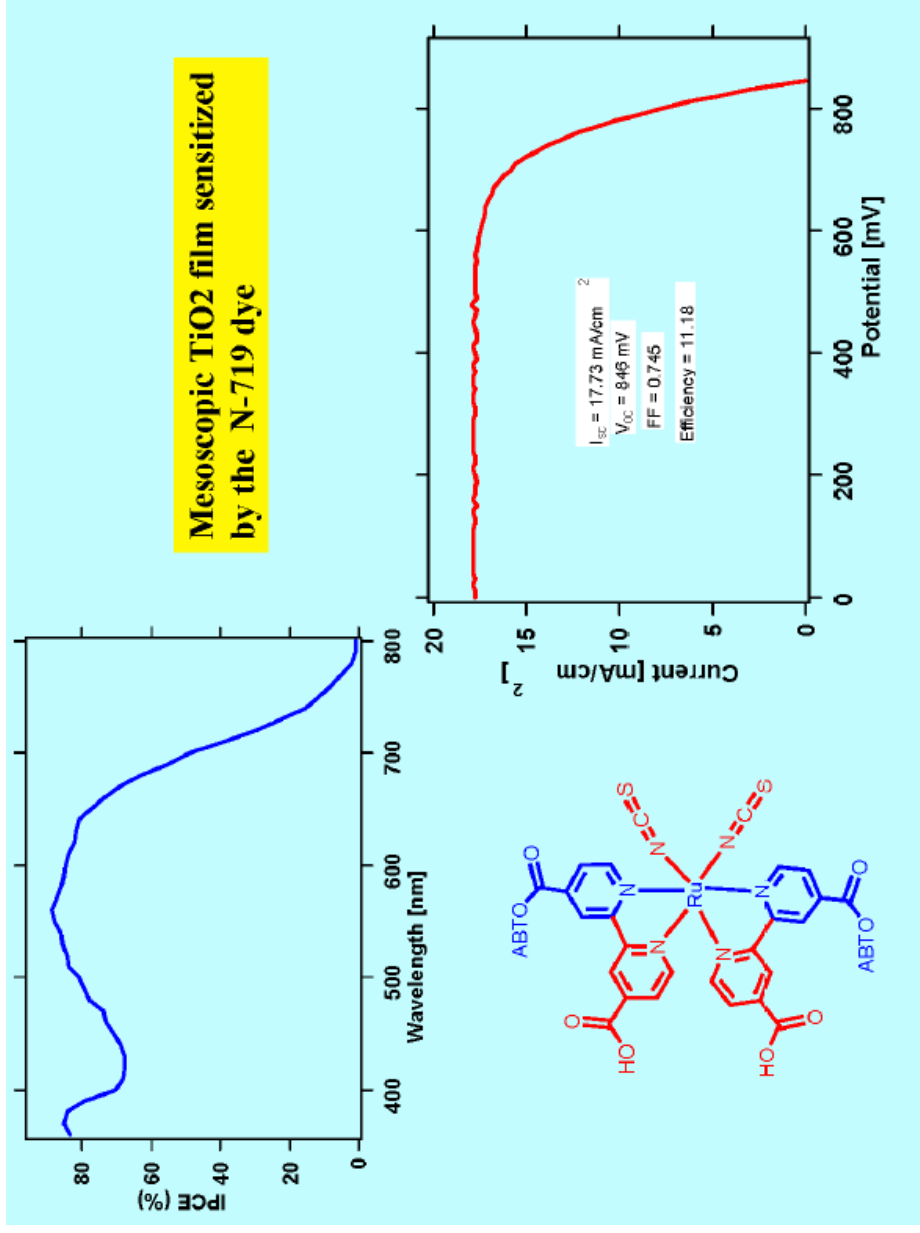
N 719 $\eta=11.2\%$



Ru oligopyridylic complexes

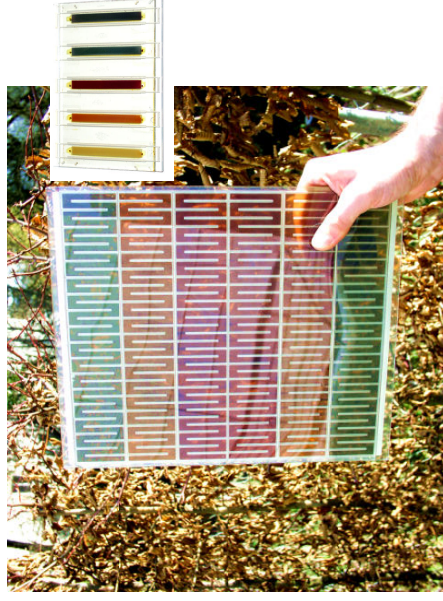
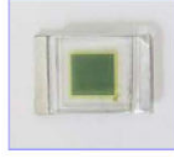
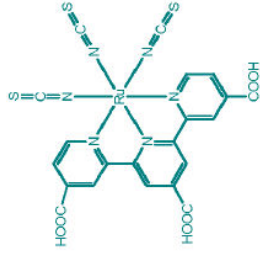
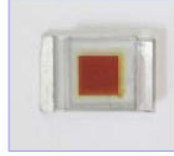
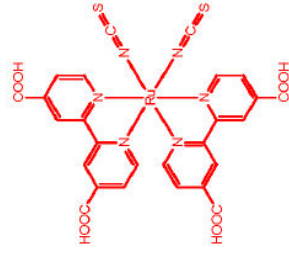
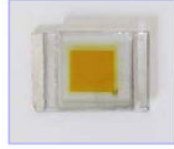
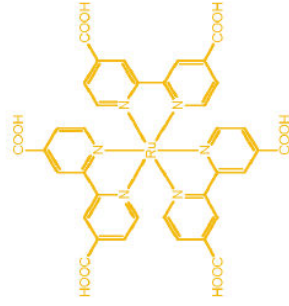


Sensitizers: Engineering Approaches



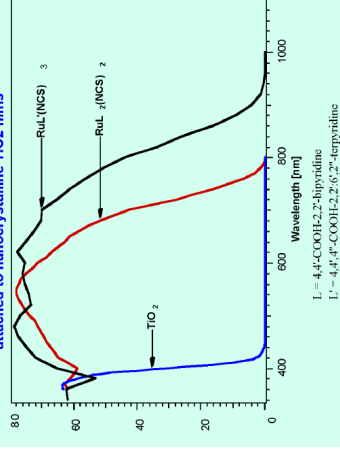
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Sensitizers: Engineering Approaches



Courtesy of RWE

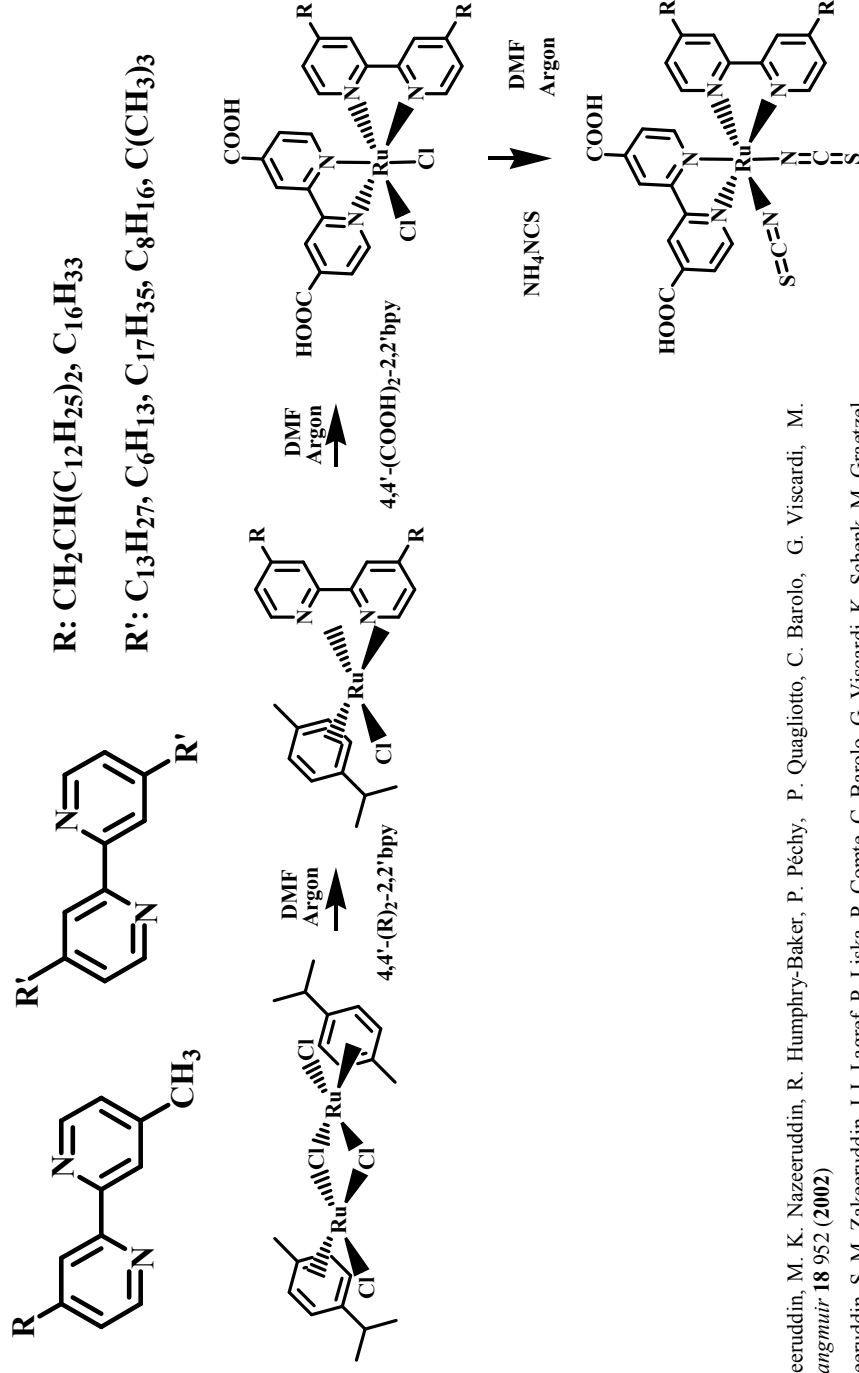
Photocurrent action spectrum of different ruthenium complexes attached to nanocrystalline TiO₂ films



L = 4,4'-COOH-2,2'-bipyridine
L' = 4,4',4''-COOH-2,2',2''-terpyridine

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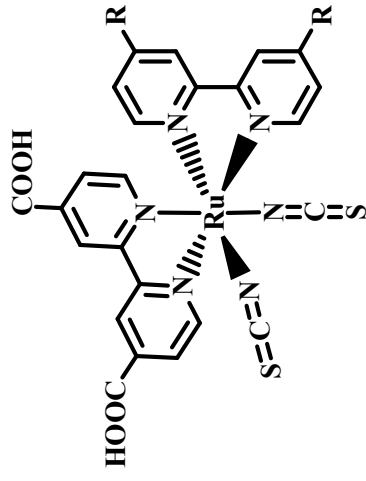
Heteroleptic Hydrophobic Sensitizers



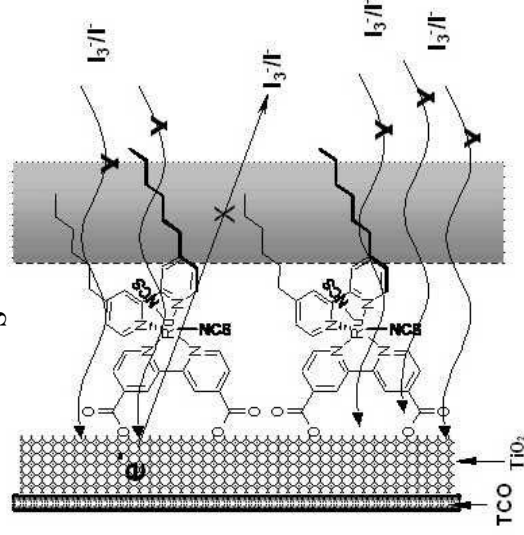
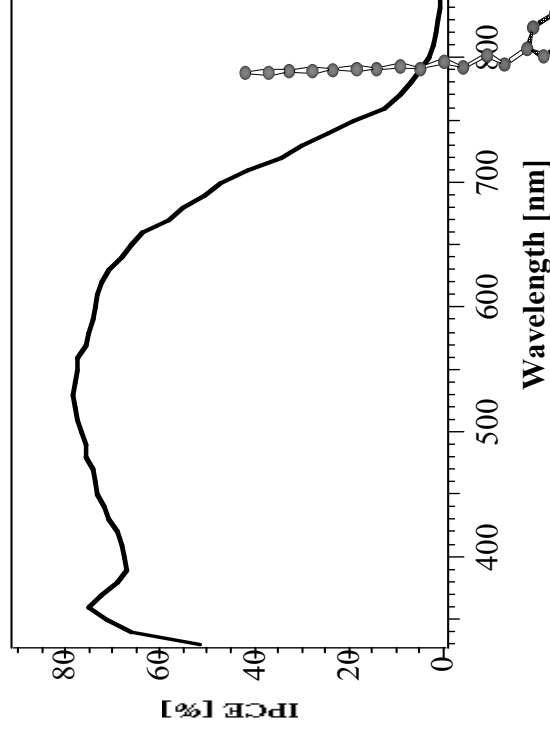
S. M. Zakeeruddin, M. K. Nazeeruddin, R. Humphry-Baker, P. Péchy, P. Quagliotto, C. Barolo, G. Viscardi, M. Graetzel *Langmuir* **18** 952 (2002)
 M. K. Nazeeruddin, S. M. Zakeeruddin, J.-J. Lagref, P. Liska, P. Comte, C. Barolo, G. Viscardi, K. Schenk, M. Graetzel *Coord. Chem. Rev.* 248 (13-14): 1317-1328 (2004)

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Heteroleptic Hydrophobic Sensitizers



- long term stability in presence of water
- better overall efficiency

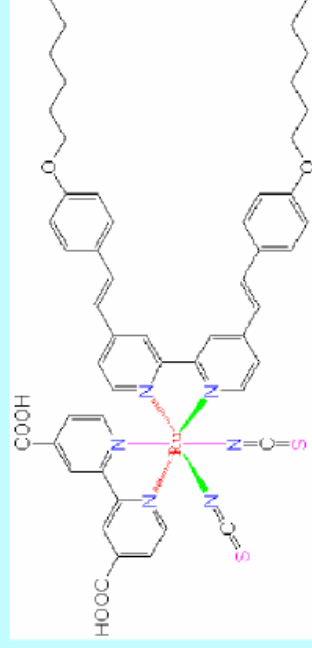


S. M. Zakeeruddin, M. K. Nazzeeruddin, R. Humphry-Baker, P. Péchy, P. Quagliotto, C. Barolo, G. Viscardi, M. Graetzel *Langmuir* **18** 952 (2002)
 M. K. Nazzeeruddin, S. M. Zakeeruddin, J.-J. Lagref, P. Liska, P. Comte, C. Barolo, G. Viscardi, K. Schenk, M. Graetzel *Coord. Chem. Rev.* **248** (13-14): 1317-1328 (2004)

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Sensitizers: Engineering Approaches

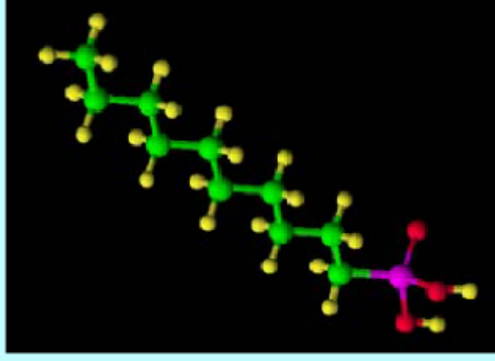
HIGH STABILITY!



K-19

Photoanode: 8+5

Decylphosphonate



ROBUST
Electrolyte

PMH: 0.8 M

I₂: 0.15 M

NMBI: 0.5 M

0.1 M GSCN

MPN solvent

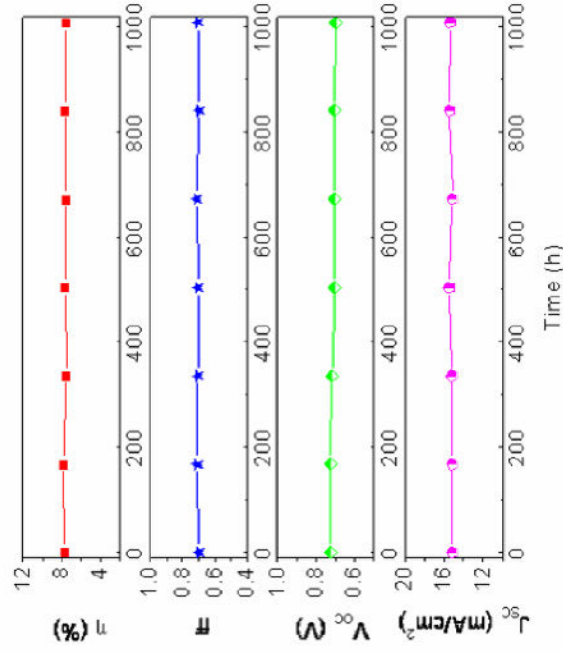
Efficiency: > 8.0%

20

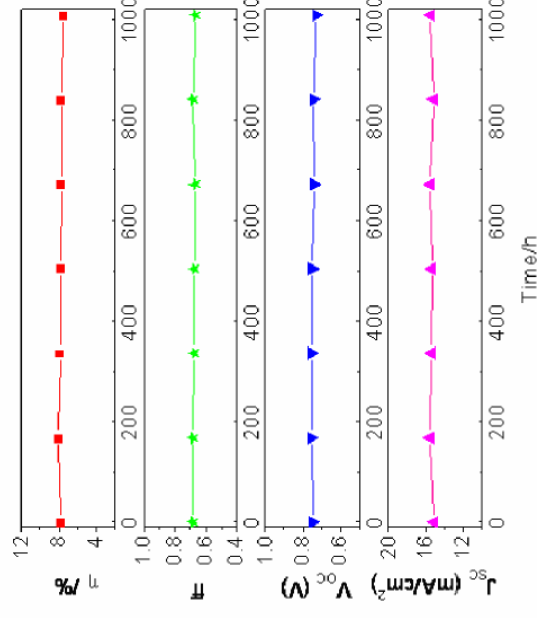
Sensitizers: Engineering Approaches

HIGH STABILITY!

80 °C evolution of device parameters in the dark



60 °C evolution of device parameters under one sun soaking



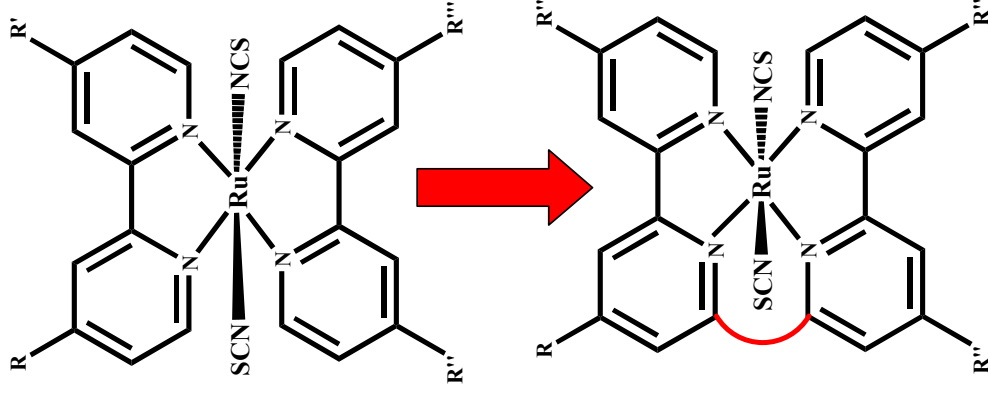
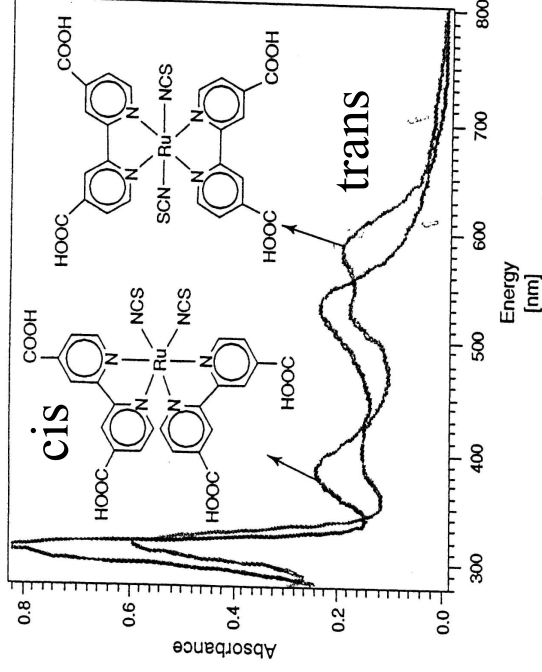
Wang, P.; Klein, C.; Humphry-Baker, R.; Zakeeruddin, S. M.; Grätzel, M. Appl. Phys. Lett. 2005, 86, 123508.

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Tetradentate Sensitizers

PURPOSE IN RESEARCH: absorption as large as possible in the visible

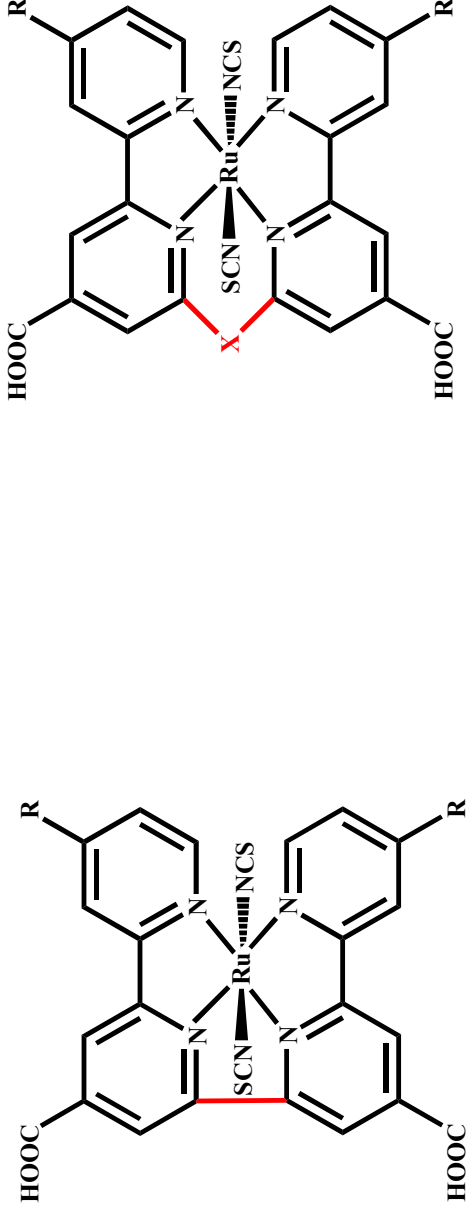
Background



Md. K. Nazeeruddin et al. *Coord. Chem. Rev.* 208 213 (2000)

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Tetradentate Sensitizers: the two models



M1

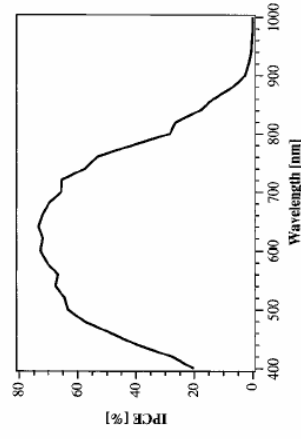
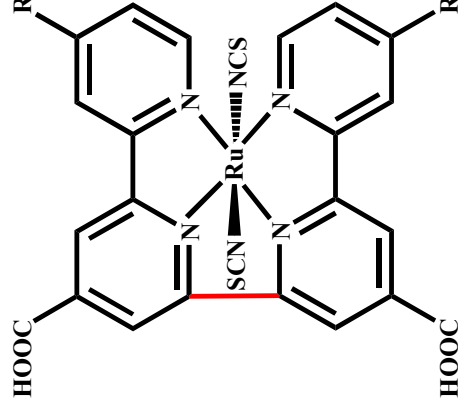
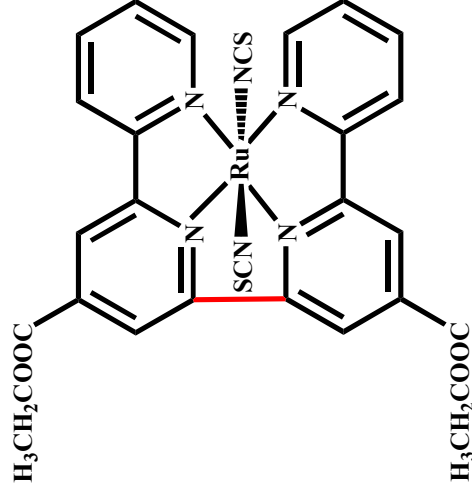
M2 and M3

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Tetradentate Sensitizers: direct linking

(M 1)



➤ Anchoring of sensitizer to TiO_2

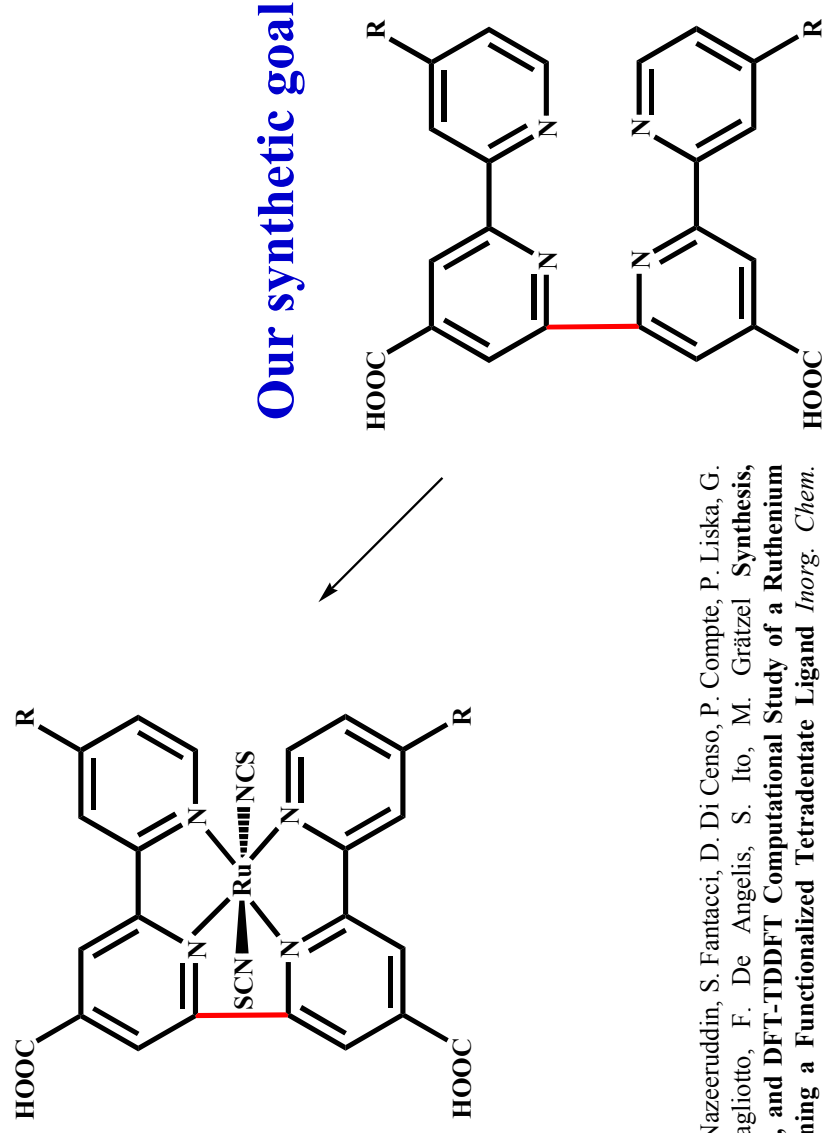
➤ Tuning of hydrophobicity to reduce: desorption
black current

M. Grätzel et al. *Tetrahedron* 57 8145 (2001)

M. Grätzel et al. *Inorganic Chemistry* 41 367 (2002)

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Tetradentate Sensitizers: direct linking (M 1)

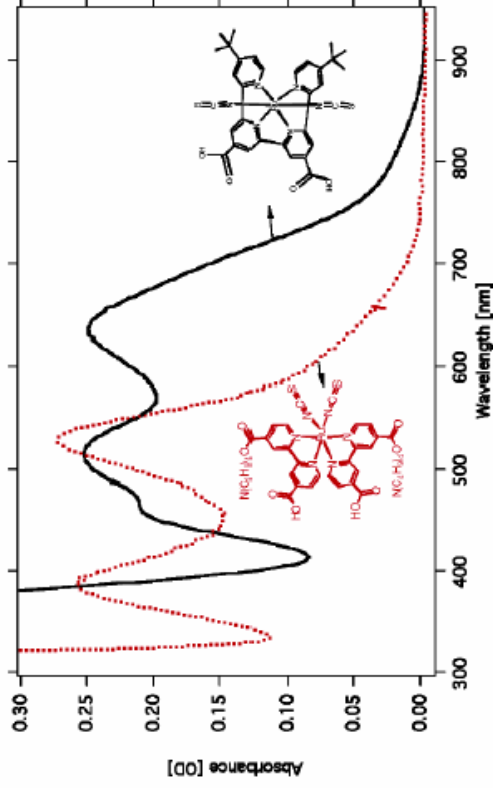
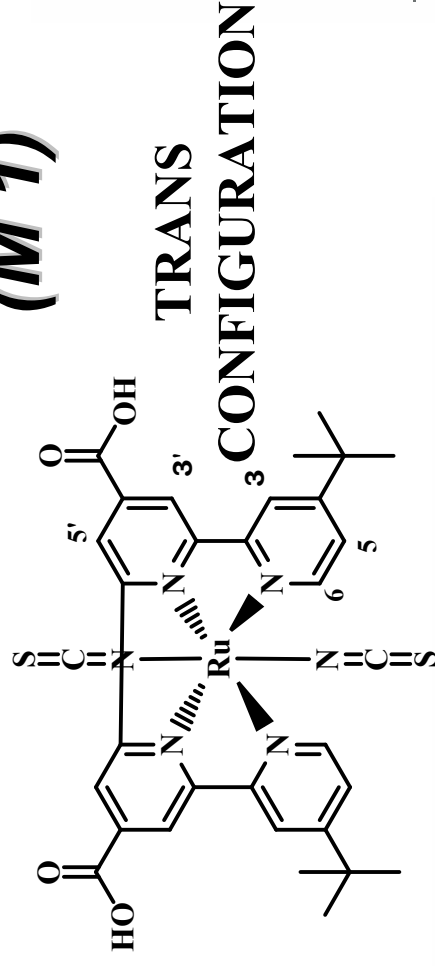


C. Barolo, M. K. Nazeeruddin, S. Fantacci, D. Di Censo, P. Compte, P. Liska, G. Viscardi, P. Quagliotto, F. De Angelis, S. Ito, M. Grätzel **Synthesis, Characterization, and DFT-TDDFT Computational Study of a Ruthenium Complex Containing a Functionalized Tetradentate Ligand** *Inorg. Chem.* (2006)

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Tetradentate Sensitizers: direct linking

(M 1)



+ 30 nm over the most bathochromic Ru sensitizer MLCT over the entire visible spectrum

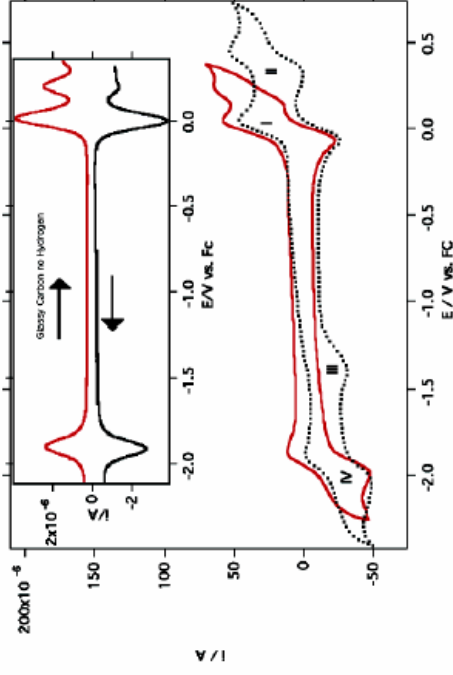
Spectral Characterization

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Tetradentate Sensitizers: direct linking

(M 1)



quasi reversible oxidation
and reduction potentials at
 $E_{1/2} = +0.38$ e -1,42 vs ferrocene,
respectively

Electrochemistry done by Davide Di Cenzo at EPFL

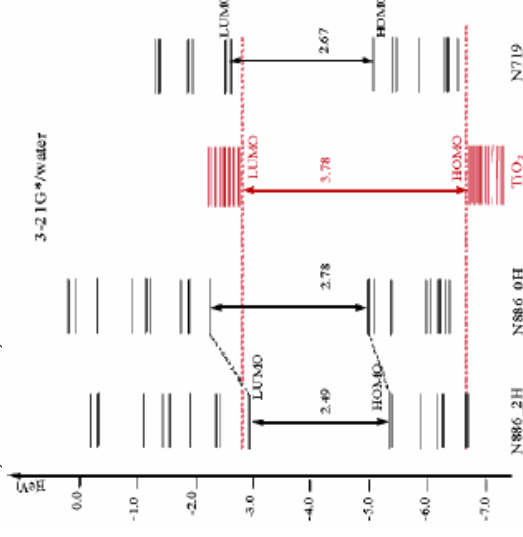
Photovoltaic data

$$V_{oc} = 680 \pm 30 \text{ mV} \text{ (-120 mV vs N786!)}$$

$$ff = 0.73 \pm 0.03$$

$$I_{sc} = 11.8 \pm 0.20 \text{ mA}$$

$$\eta = I_{sc} V_{oc} ff = 5.85 \%$$



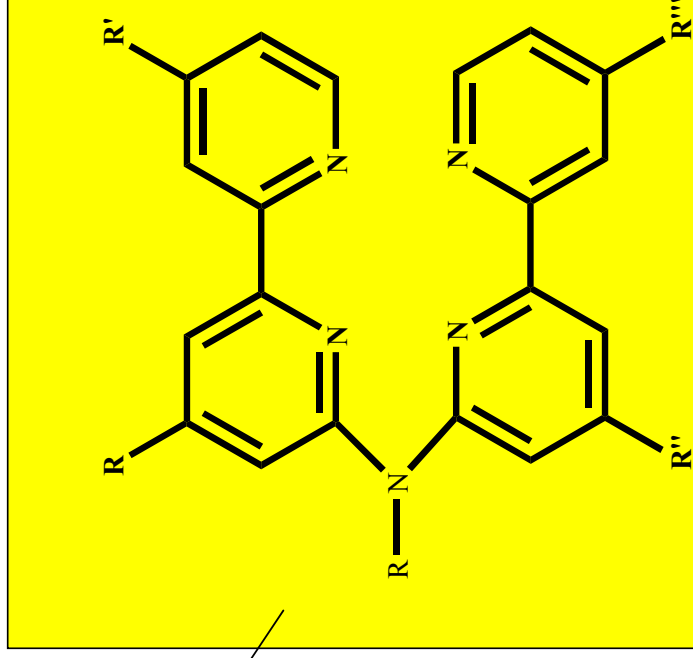
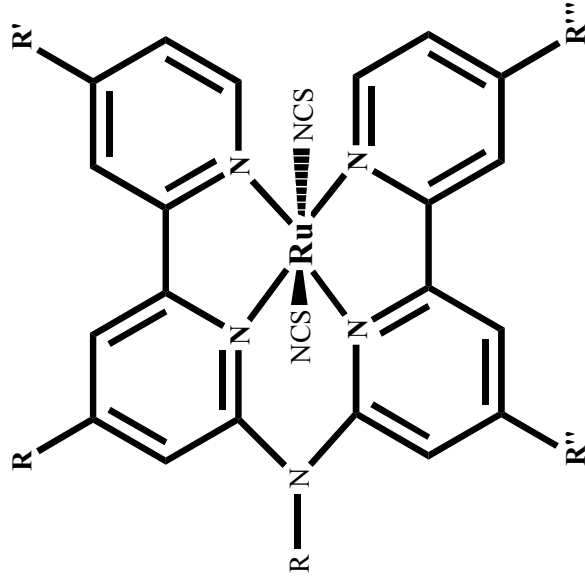
DFT-TDDFT calculations done by
Fantacci and De Angelis, Perugia

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Tetradentate Sensitizers: amino bridge

(M 2)

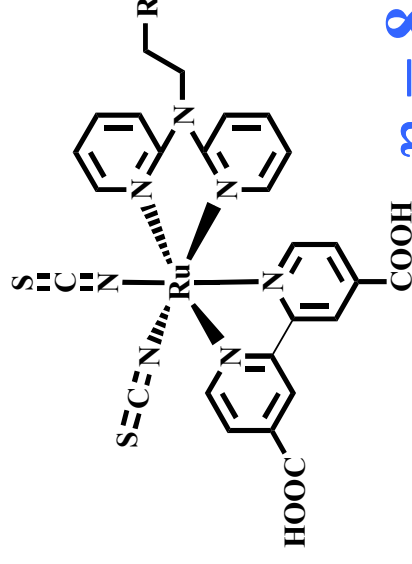
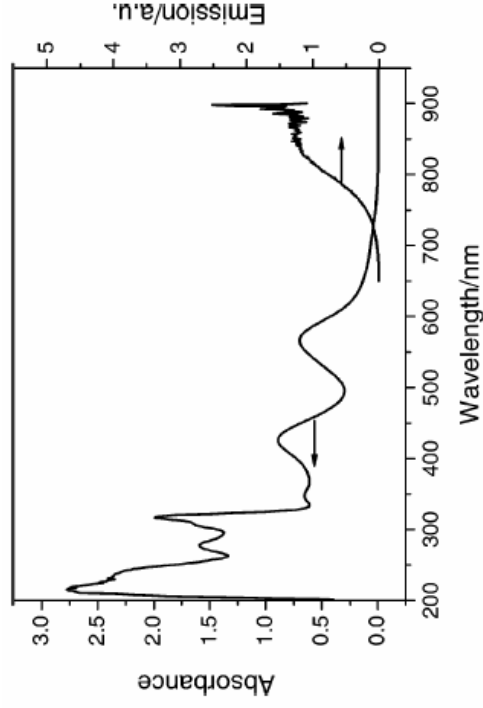


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Tetradentate Sensitizers: amino bridge (M 2)

Graetzel et al. *Chem.Mater.* 2004 16, 3246-3251



$\eta = 8.2\%$

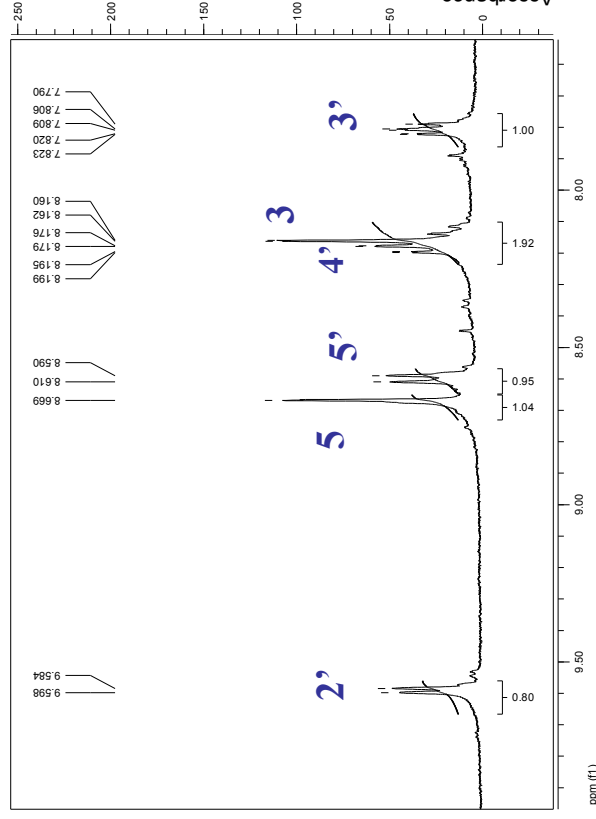
Table 1. Detailed Photovoltaic Parameters of DSCs with Complex I under Different Incident Light Intensities^a

$P_{in}/mW\ cm^{-2}$	$J_{sc}/mA\ cm^{-2}$	V_{oc}/mV	$P_{max}/mW\ cm^{-2}$	ff	$\eta/\%$
9.5	1.54	692	0.84	0.786	8.8
52.1	8.38	740	4.53	0.745	8.7
99.7	15.5	756	8.21	0.702	8.2

^a The spectral distribution of the lamp simulates air mass 1.5 solar light. Incident power intensity: P_{in} ; short-circuit photocurrent density: J_{sc} ; open-circuit photovoltage: V_{oc} ; maximum electricity output power density: P_{max} ; fill factor: $ff = P_{max}/P_{in}$; total power conversion efficiency: η ; cell active area: $0.158\ cm^2$.

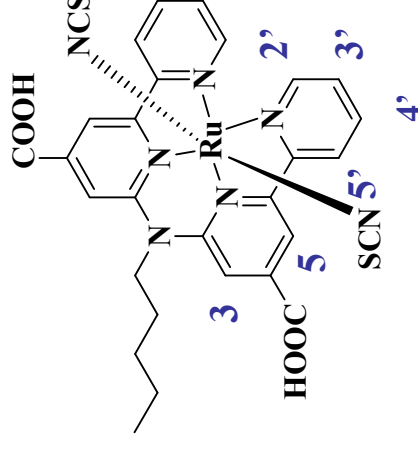
Tetradentate Sensitizers: amino bridge

(M 2)



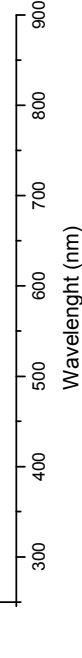
¹H-NMR:

- 7.81 ppm (2H, dd, H3'')
- 8.20 ppm (2H, dd, H4'')
- 8.16 ppm (2H, s, H3)
- 8.60 ppm (2H, d, H5'')
- 8.67 ppm (2H, s, H5)
- 9.59 ppm (2H, d, H2'')



$$\eta = 2.79 \%$$

red-violet dye



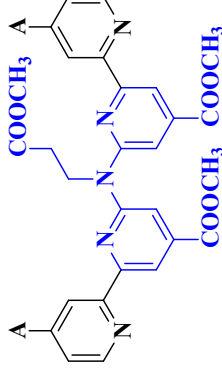
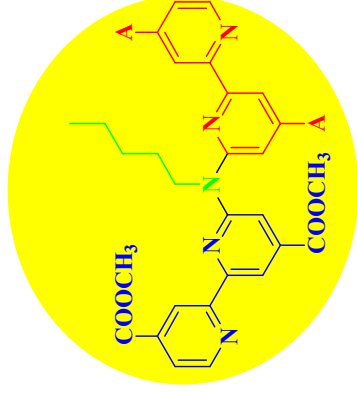
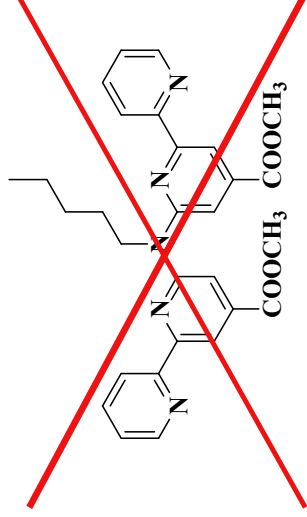
TRANS CONFIGURATION

C. Barolo – Dye Sensitized Solar Cells

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Tetradentate Sensitizers: amino bridge

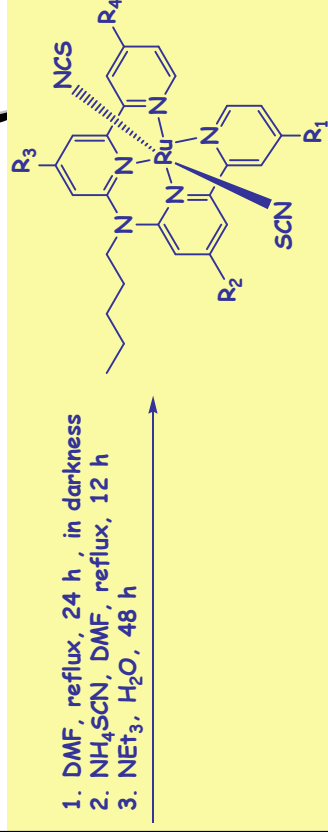
(M 2)



C. Barolo – Dye Sensitized Solar Cells

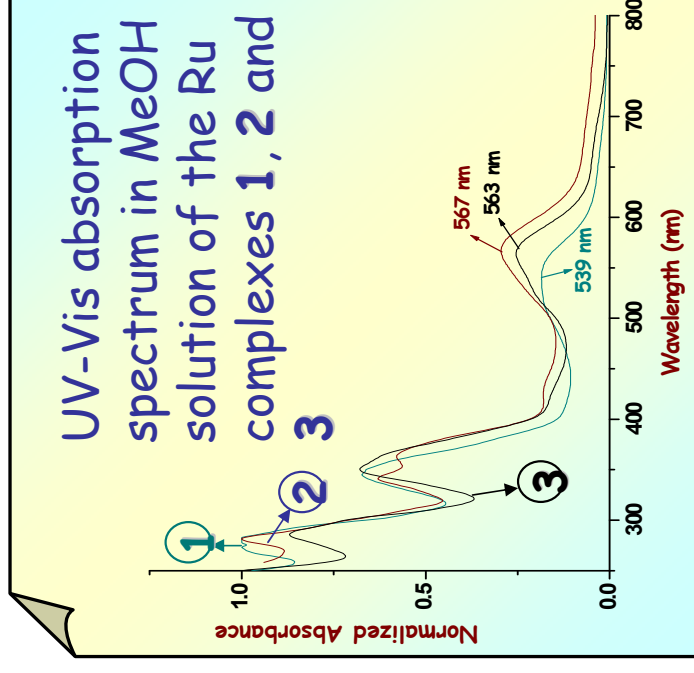
Tetradentate Sensitizers: amino bridge

(M 2)



Complex 1: 2 COOH centered $\eta = 2.79 \%$

Complex 2: 2 COOH one side $\eta = 4.30 \%$



Complex 3: 4 COOH

$\eta = 7.22 \%$



C. Barolo – Dye Sensitized Solar Cells

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Organic dyes

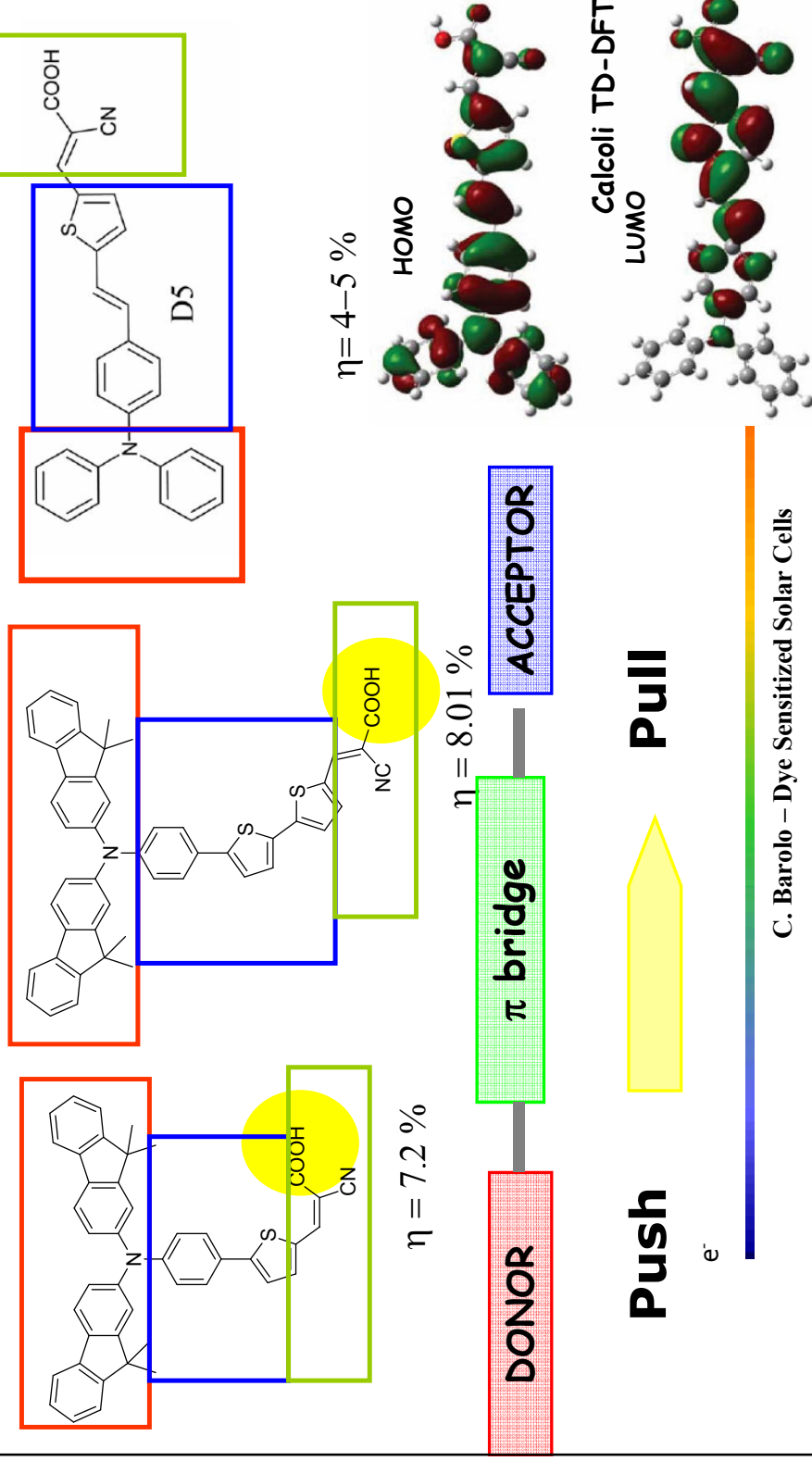
Background in solar cell

Metal free sensitizers in DSSC's since 2003

Efficiency range: 5-8%

Hagfeldt et al. *Chem. Commun.* 2006 2245-2247

Graetzel et al. *JACS* 2006 128 16701-16707



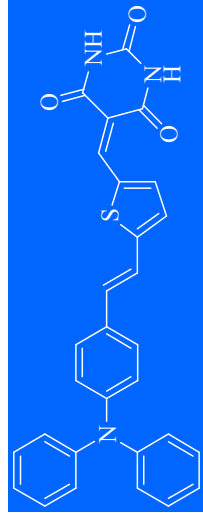
Organic dyes



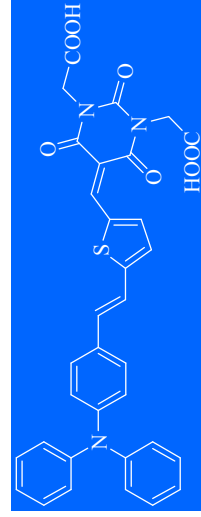
η @1Sun	J_{sc} (mA/cm ²)	V_{oc} (mV)	ff
5.55 %	11.21	664.2	0.75



η @1Sun	J_{sc} (mA/cm ²)	V_{oc} (mV)	ff
4.5 %	9.83	636.38	0.72



η @1Sun	J_{sc} (mA/cm ²)	V_{oc} (mV)	ff
0.81 %	2.25	480.34	0.75



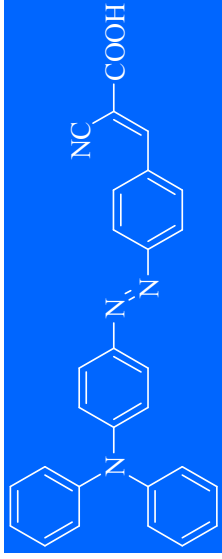
η @1Sun	J_{sc} (mA/cm ²)	V_{oc} (mV)	ff
0.15 %	0.44	466.22	0.75

AGGREGATION!



C. Barolo – Dye Sensitized Solar Cells

Organic dyes



$\eta @ 1$ Sun	J_{sc} (mA/cm ²)	V_{oc} (mV)	FF
0,35%	0,83	549,06	0,78

- No aggregation problems!
- LUMO at the right level (from CV point of view)
- excitation state life-time too short (azo isomerization process?)

NOT ONLY SYNTHESIS ...

PRODUCT PURIFICATION

C. Barolo – Dye Sensitized Solar Cells

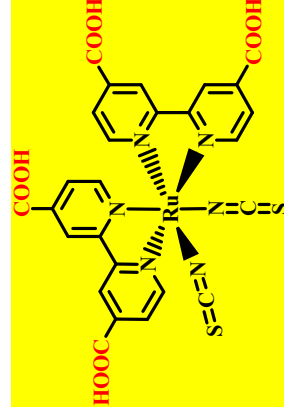
35

NOT ONLY SYNTHESIS ...

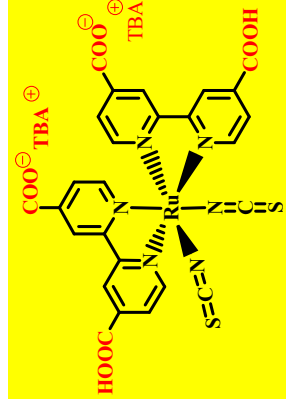
PRODUCT CHARACTERISATION

A Mass Spectrometric Analysis of N-719 Sensitizer Solution used for Dye-Sensitized Solar Cell

WHY?



N3



N719

The high performance achieved using (N719) sensitized TiO₂ solar cell can be related to several factors:

- broad range of visible light absorption,
- relatively long-lived excited states,
- its excited energy almost matching those of TiO₂ conduction band
- **the effect of protons and the cations carried by the sensitizer**

G. Viscardi et al. *Inorganica Chimica Acta*, 2008

C. Barolo – Dye Sensitized Solar Cells

NOT ONLY SYNTHESIS ...

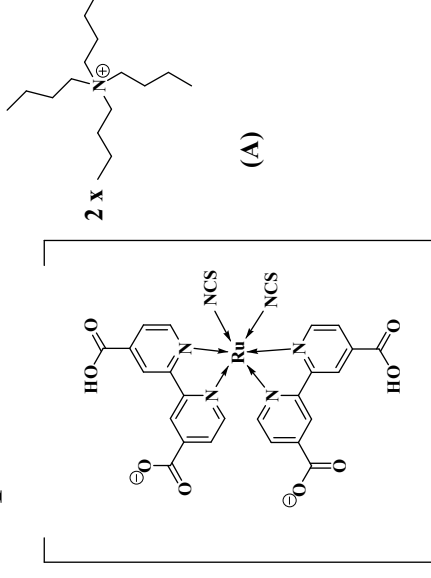
PRODUCT CHARACTERISATION

First Mass Spectrometric Study with a ESI interface



ANALYZE THE SENSITIZER
STRUCTURE DIRECTLY FROM
ADSORBING SOLUTION in order to
design and develop new tailored sensitizers

- ✓ it can directly detect mono and multi-charged ion
- ✓ does not alter the connectivity of the complex through disassembly/reassembly sequences
- ✓ causes essentially no or very little fragmentation.



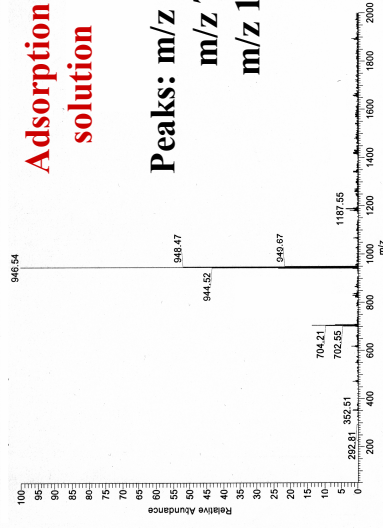
N719
 $C_{58}H_{86}N_8O_8RuS_2$
Exact Mass: 1188,51

C. Barolo – Dye Sensitized Solar Cells

NOT ONLY SYNTHESIS ...

PRODUCT CHARACTERISATION

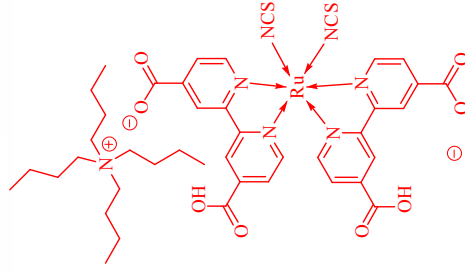
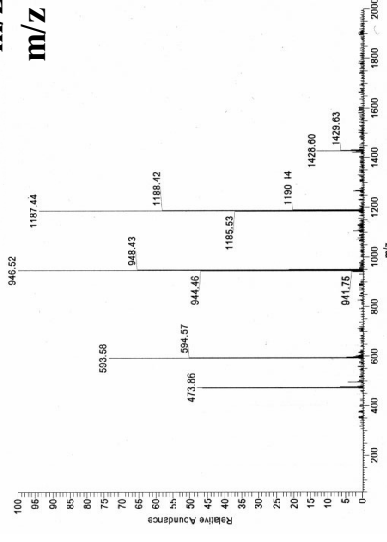
Adsorption solution



Solvent effect

H₂O+5%CH₃OH

Peaks: m/z 1187,5 (F)
m/z 593,2 (D)
m/z 473,9 (C)
m/z 1428,8 (G)



m/z 946,1

C. Barolo – Dye Sensitized Solar Cells

DSC INDUSTRIALIZATION

Advantages vs. Silicon Cells

- No feedstock supply problems, low cost and ease of production,
- Performance insensitive to temperature
- Bifacial configuration - advantage for diffuse light and albedo
- Efficiency less sensitive to angle of incidence,
- 10 -30 percent higher energy output than silicon cells at equal SRC rating
- Transparency for power windows
- Color can be varied by selection of the dye, invisible PV-cells based on near-IR sensitizers are feasible
- Low energy content (for silicon this is 5 GJ/m² !), payback time is only a few months as compared to years for silicon.
- Outperforms single junction amorphous Si

DSC INDUSTRIALIZATION



INDUSTRIALISATION OF DSC FROM RESEARCH TO PRODUCT 9 & 10 FEBRUARY 2006

INVITATION

The inaugural conference "Industrialisation of DSC—from Research to Product" is being held in Canberra, Australia 9 & 10 February 2006.



Organiser



Co-Organiser

2nd INTERNATIONAL CONFERENCE on the INDUSTRIALISATION of DSC

DSC-IC 2007
St Gallen, Switzerland, 11-13 September 2007

Program Overview Speakers Registration DSC Network Sponsors Hotel and Travel See & Do

DSC INDUSTRIALIZATION

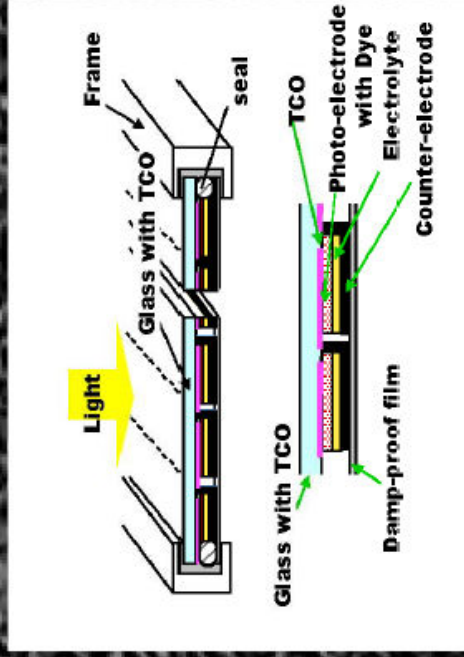
The Toyota Dream House



DSC INDUSTRIALIZATION

AISIN Prototype production

- Search of new module design from industrial point of view < performance, durability, number of parts, production time, cost >



DSC INDUSTRIALIZATION

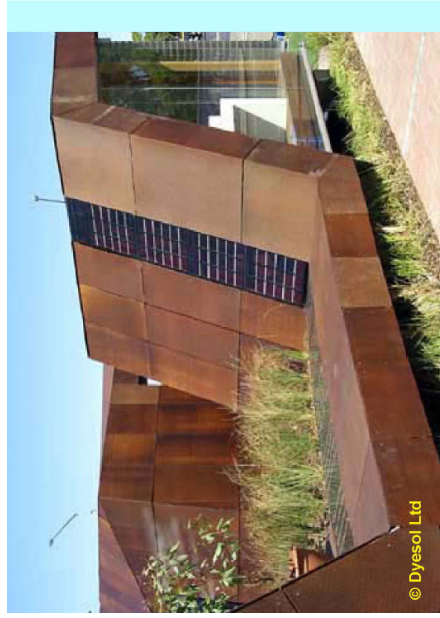
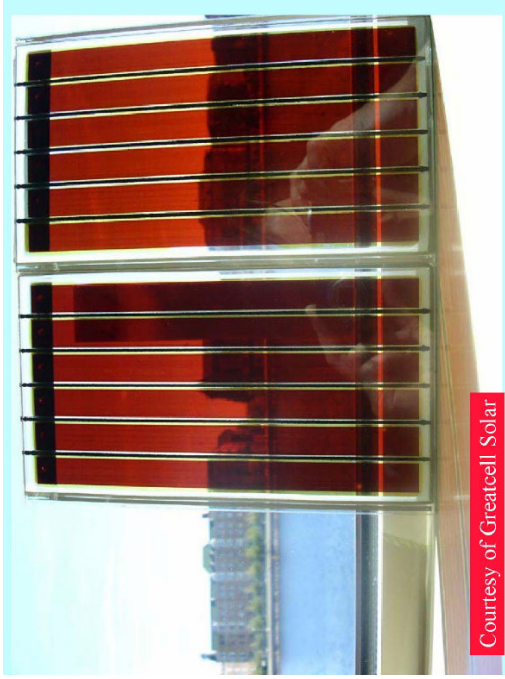


Hitachi's new dye sensitized cell achieves 9.3 percent efficiency

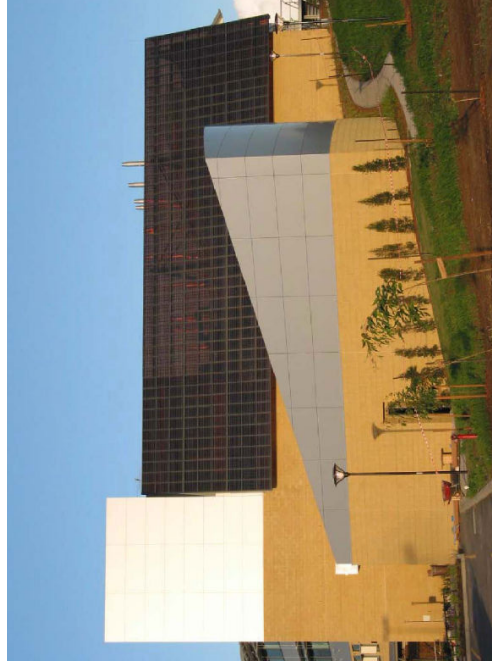


C. Barolo – Dye Sensitized Solar Cells

DSC INDUSTRIALIZATION



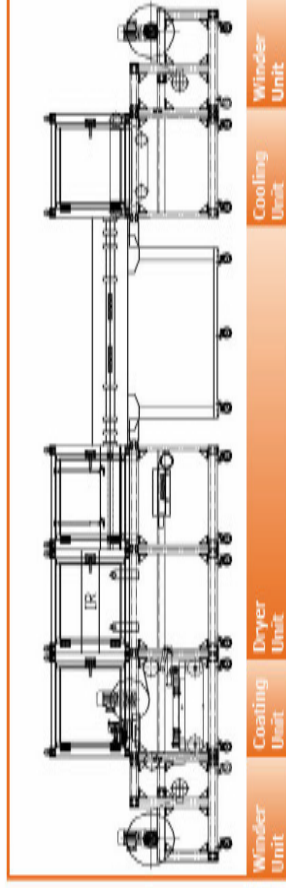
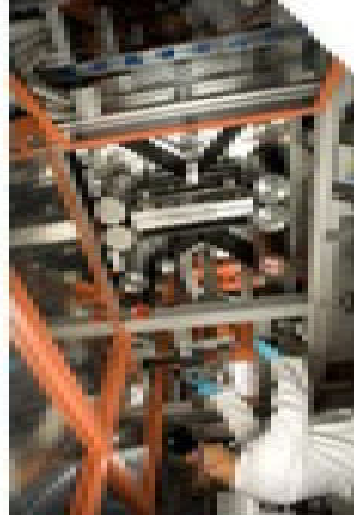
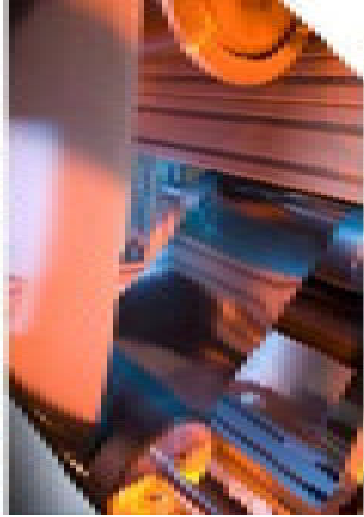
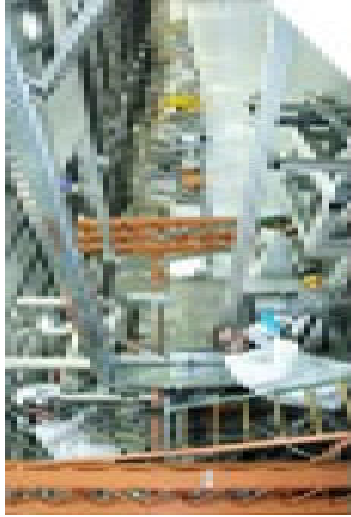
10 m² of Dyesol DSC facade panels have been integrated to form a magenta «stripe» across the undulating wall floor-roof of one of the Houses of the Future on display at the Sydney Olympic Park.



C. Barolo – Dye Sensitized Solar Cells

DSC INDUSTRIALIZATION

G24i
First production plant: Cardiff, Galles
roll to roll



C. Barolo – Dye Sensitized Solar Cells

ACKNOWLEDGMENTS

Prof. Ermanno Barni and Prof. Guido Viscardi
Chimica Organica Applicata



Prof. Michael Graetzel
LPI group at EPFL



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We thank Compagnia di San Paolo and Fondazione CRT, Torino Italy for continuous equipment supplied.

AND NOW?

C. Barolo – Dye Sensitized Solar Cells

NIS Colloquium

Classical and new approaches to thin film photovoltaics

Monday, June 23, 2008

Aula Magna, Experimental Physics Department, Via Pietro Giuria 1, Torino

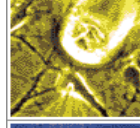
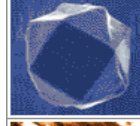
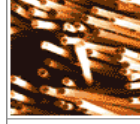
DYE SENSITIZED SOLAR CELLS

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Dipartimento di Chimica Generale e Chimica Organica
Centro di Eccellenza Interdipartimentale NIS (Superfici ed Interfasi Nanostrutturate)
Università degli Studi di Torino

THANK YOU FOR YOUR ATTENTION !



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